

Chapter 11 Review Gases Section 1 Worksheet Answers

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Chapter 11 Review Gases Section

SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c) ripld. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOictbv-x

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SECTION 3 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (b) divided by the mass of 1 mol nRT (c) multiplied by 224 L (d) divided by 224 L For the expression $V =$ (a) increasing P

[MOBI] Chapter 11 Review Gases Section 2

CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided by the mass of 1 mol (d) divided by 224 L 2 For the expression $V = nR/P$

[DOC] Chapter 11 Review Gases Section 2 Answers

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Chapter 11 Review Gases Section 1 Answers

Modern Chemistry 97 Gases CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. ____ For the expression , P

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CHAPTER 11 REVIEW . Gases . SHORT ANSWER . Answer the followin9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of l mol. (c) multiplied by 22.4 . L. (b) divided by the mass of 1 mol. (d) divided by 22.4 . L. 2. c For the expression $V = n\sim T,$

CHAPTER REVIEW Gases

CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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CHAPTER 11 REVIEW. Gases. SECTION 3. SHORT ANSWER Answer the following questions in the space provided. 1. The molar mass of a gas at STP is the density of that gas. (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2.

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Chapter 11 - Gases Chapter 11 focuses on gas behavior and the gas laws. In Chapter 10, students were given an overview of the kinetic-molecular theory of matter and discussed how this theory...

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CHAPTER 11 REVIEW. Gases. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. ____ For a constant force, when the surface area is tripled the pressure is. (a) doubled. (b) a third as much.

01 - Leon County Schools

Chapter 11 Gases. Chapter 11. Gases. 459. t's Monday morning, and Lilia is walking out of the chemistry building, thinking about the introductory lecture on gases that her instructor just presented. Dr.

Chapter 11 Gases - An Introduction to Chemistry

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