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5070/2 Jun02 [Turn over A2 Hydrogen reacts with chlorine to

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make hydrogen chloride. The reaction is exothermic. The reaction can be represented by the equation below.  $H_2 + Cl_2 \rightarrow 2HCl$   $\Delta H = -184 \text{ kJ/mol}$  (a) A mixture of 2.5g of hydrogen and 142g of chlorine is allowed to react.

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We have increased the number of marks and paper duration in Paper 2 and removed Section B (optional questions). We have decreased the number of marks in Paper 4. This paper now has a similar structure to Paper 3, making sure there is consistency between the two papers that assess practical skills, including planning questions.

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## **Cambridge O Level Chemistry (5070)**

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